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pure water placed in contact with solid  $\text{AgCl}(s)$ ?  $K_{sp} = [\text{Ag}^+] [\text{Cl}^-]$   
 $1.8 \times 10^{-10} = [x] [x]$   $x = 0.000013416 \text{ M}$  or  $1.3416 \times 10^{-5} \text{ M}$  2) The  
molar solubility of  $\text{PbCl}_2$  ...

## *Ksp Problems - Chemistry*

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3-3 Practice Problems 1. How many protons and electrons are 8. What is  
thename of element that has present in a vanadium atom? atoms that  
contain 25 protons? 2. How many protons and electrons are 9. What is  
the name of the element that has present in a nitrogen atom? atoms  
that contain 82 protons? 3. How many protons and electrons are 10.

## *Name Date Class 3-3 Practice Problems - Weebly*

0.710 kg of sodium carbonate (washing soda),  $\text{Na}_2\text{CO}_3$ , in 10.0 kg of  
water—a saturated solution at  $0^\circ\text{C}$ ; 125 g of  $\text{NH}_4\text{NO}_3$  in 275 g of  
water—a mixture used to make an instant ice pack; 25 g of  $\text{Cl}_2$  in 125

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g of dichloromethane,  $\text{CH}_2\text{Cl}_2$ ; 0.372 g of histamine,  $\text{C}_5\text{H}_9\text{N}$ , in 125 g of chloroform,  $\text{CHCl}_3$

*8.3: Concentrations of Solutions (Problems) - Chemistry ...*

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