

## Covalent Compounds Worksheet Answers

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Covalent Compound Naming Worksheet 1covalentname.sxw. Naming Covalent Compounds Solutions. Write the formulas for the following covalent compounds: 1) antimony tribromide SbBr<sub>3</sub>. 2) hexaboron silicide B<sub>6</sub>Si 3) chlorine dioxide ClO<sub>2</sub>. 4) hydrogen iodide HI 5) iodine pentafluoride IF<sub>5</sub>. 6) dinitrogen trioxide N<sub>2</sub>O<sub>3</sub>.

[Covalent Compound Naming Worksheet - Prairie Science](#)

Naming Covalent Compounds Worksheet PART I: Write the formulas for the following covalent compounds: antimony tribromide 2. 3. 4. 5. 6. 8. 9. 10. 11. 12. 13. 14. 15. 16. 17. 18. 19. 20. 21. 22. 23. hexaboron silicide chlorine dioxide hydrogen iodide iodine pentafluoride dinitrogen trioxide phosphorus triiodide Phosphorus trihydride Dichlorine heptoxide

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Ionic And Covalent Bonding Worksheet Answers. Determine if the elements in the following compounds are metals or non metals describe the type of bonding that occurs in the compound. It may therefore be presumed this might be a reason as to why ionic and covalent substances arent excellent conductors of electricity.

[Ionic And Covalent Bonding Worksheet Answers](#)

Ionic/Covalent Compound Naming Solutions. For each of the following questions, determine whether the compound is ionic or covalent and name it appropriately. 1) Na<sub>2</sub>CO<sub>3</sub> sodium carbonate. 2) P<sub>2</sub>O<sub>5</sub> diphosphorus pentoxide. 3) NH<sub>3</sub> ammonia. 4) FeSO<sub>4</sub> iron (II) sulfate

[Mixed ionic and covalent naming worksheet and answers ...](#)

The answer to the worksheet you are doing is covalent bonds. What is the covalent compound of SCl<sub>6</sub>? This is a covalent compound. S-Cl bond is covalent. Dichlorobenzene ionic or covalent compound ...

[Covalent compound worksheet? - Answers](#)

Nomenclature Worksheet 3 – Covalent (Molecular) Compounds Part A: Name the following covalent compounds. 1. CO Carbon monoxide . 2. CO<sub>2</sub> Carbon dioxide. 3. N<sub>2</sub>O<sub>3</sub> Dinitrogen trioxide. 4. N<sub>2</sub> omit. 5. NP Nitrogen monophosphide. 6. SCl<sub>2</sub> Sulfur dichloride. 7. P<sub>2</sub>O<sub>5</sub> Diphosphorous pentoxide. 8. NBr<sub>3</sub> Nitrogen tribromide. 9. Cl<sub>4</sub> Carbon tetraiodide. 10.

[Nomenclature Worksheet 3 – Covalent \(Molecular\) Compounds](#)

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Chemistry Worksheet Lewis Dot Structures Name: Block. 2. Draw Lewis structures for the following covalent compounds: H<sub>2</sub>O<sub>2</sub>. C<sub>2</sub>H<sub>5</sub>Cl CH<sub>3</sub>COCH<sub>3</sub> c- CH<sub>2</sub>O (carbon is central) CO<sub>2</sub> (carbon is central) C<sub>2</sub>H<sub>5</sub>OH CH<sub>3</sub>NH<sub>2</sub> H c Page 2 of 2 Copyright © 2007, Newton South High School, Newton, MA 02459. Created Date. 3/28/2014 11:29:01 AM.

[Chemistry Worksheet Lewis Dot Structures Name: Block: 1 ...](#)

Covalent Bond between a Non-Metal and Non-Metal (NM + NM) Determine if the elements in the following compounds are metals or non-metals. Describe the type of bonding that occurs in the compound. Compound Element 1 (metal or non-metal?) Element 2 (metal or non-metal?) Bond Type NO 2 N = non-metal O = non-metal covalent NaCl M NM Ionic SO<sub>2</sub> NM Covalent PI

[WORKSHEET: Ionic vs. Covalent! REMEMBE Ionic Bond Covalent ...](#)

Answers – Naming Chemical Compounds Name the following chemical compounds: 1) NaBr sodium bromide 2) Ca(C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>)<sub>2</sub> calcium acetate 3) P<sub>2</sub>O<sub>5</sub> diphosphorus pentoxide 4) Ti(SO<sub>4</sub>)<sub>2</sub> titanium(IV) sulfate 5) FePO<sub>4</sub> iron(III) phosphate 6) K<sub>3</sub>N potassium nitride 7) SO<sub>2</sub> sulfur dioxide 8) CuOH copper(I) hydroxide 9) Zn(NO<sub>2</sub>)<sub>2</sub> zinc nitrite 10) V<sub>2</sub>S<sub>3</sub> vanadium(III) sulfide

*Chemistry: Naming Covalent Compounds Worksheet Name Write ...*

Answers – Naming Chemical Compounds . Name the following chemical compounds: 1) NaBr sodium bromide. 2) Ca(C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>)<sub>2</sub> calcium acetate. 3) P<sub>2</sub>O<sub>5</sub> diphosphorus pentoxide. 4) Ti(SO<sub>4</sub>)<sub>2</sub> titanium(IV) sulfate. 5) FePO<sub>4</sub> iron(III) phosphate. 6) K<sub>3</sub>N potassium nitride. 7) SO<sub>2</sub> sulfur dioxide. 8) CuOH copper(I) hydroxide. 9) Zn(NO<sub>2</sub>)<sub>2</sub> zinc nitrite

*Answers – Naming Chemical Compounds*

Worksheet Chemical Bonding Ionic and Covalent Answers Part 2 Along with Lovely Ionic Bonding Worksheet Answers Best Chemical Bonds. The fact that a molecule will react when it has an ionic bond is not new to us.

*Worksheet Chemical Bonding Ionic and Covalent Answers Part 2*

correct naming rules to write the correct names for each compound. Chemical Formula Type of Compound: Ionic or Covalent Compound Name 21) CdBr<sub>2</sub> 22) Cr<sub>2</sub>O<sub>7</sub> 23) SBr<sub>2</sub> 24) (NH<sub>4</sub>)<sub>2</sub>CrO<sub>4</sub> 25) CuO 26) Pt<sub>3</sub>(PO<sub>3</sub>)<sub>4</sub> 27) Al(ClO<sub>4</sub>)<sub>3</sub> 28) NH<sub>3</sub> 29) Ca(C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>)<sub>2</sub> 30) N<sub>2</sub>O Chemical Formula Type of Compound: Ionic or Covalent Compound Name 31) V(SO<sub>4</sub>)<sub>2</sub> 32) Ag<sub>2</sub>CO<sub>3</sub> 33) N<sub>2</sub>S

*Ionic & Covalent Compound Naming Race*

Naming Mixed Ionic and Covalent - Answers Name the following compounds. Remember, they may be either ionic or covalent compounds, so make sure you use the right naming method! 1) NaF sodium fluoride 2) NF<sub>3</sub> nitrogen trifluoride 3) Li<sub>2</sub>O lithium oxide 4) Al<sub>2</sub>S<sub>3</sub> aluminum sulfide 5) MgSO<sub>4</sub> magnesium sulfate 6) SiH<sub>4</sub> silicon tetrahydride 7) KNO<sub>3</sub> potassium nitrate

*Naming Ionic Compounds – Answer Key*

Thus, the compound formed from sodium and chlorine will be ionic (a metal and a non-metal). Nitrogen monoxide (NO) will be a covalently bound molecule (two non-metals), silicon dioxide (SiO<sub>2</sub>) will be a covalently bound molecule (a semi-metal and a non-metal) and MgCl<sub>2</sub> will be ionic (a metal and a non-metal).

*4.2: Covalent Compounds - Formulas and Names - Chemistry ...*

Many of the compounds in your body are covalent compounds. Covalent compounds form when atoms share electrons. The bond that forms when atoms share electrons is called a covalent bond. Atoms share electrons to fill their outermost energy level. This forms a group of atoms, each having a full valence shell. The group of atoms that make up a covalent compound is called a molecule. A molecule is the smallest particle

*3 SECTION 1 Ionic and Covalent Compounds*

12 Best Images Of Naming Covalent Compounds Worksheet Worksheets are covalent bonding work with answers, university of texas at austin, ionic bond middle school work, chemical bonds ionic bonds answers, ionic bonds work answers, chapters 6 and 7 practice work covalent bonds and, covalent, atomic structure. click on pop out icon or print icon to worksheet to print or download.

*Ionic And Covalent Bonding Practice Worksheet Answers ...*

After you complete a step (or answer a question), place a check mark in the box (✓) next to that step. ... Properties of Ionic and Covalent Compounds. Student Inquiry Worksheet. 2 PS-2871C 3 Name Period Date. 1 Identify the bonding characteristics of the unknown compounds. After testing the known compounds, test the physical properties of each ...

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