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set (32) ... In each case,  
the single intermediate is  
the ES complex. (see page  
190 or picture from test  
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strict "lock and key" model,  
whereas (b) describes a  
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Ethanol in Water Explain why  
ethanol ( $\text{CH}_3\text{CH}_2\text{OH}$ ) is  
more soluble in water than  
is ethane ( $\text{CH}_3\text{CH}_3$ ).

Answer Ethanol is polar;

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