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Name CHAPTER 3 TEST continued Date Class FILL IN THE BLANK Write the correct term (or terms) in the space provided. 9. If a particular compound is composed of elements A and B, the ratio of the mass of B to the mass of A will always be the same. This is a statement of the law of exactly 12 g of carbon-12 is referred to as a(n) 11.

## ~~San Ramon Valley High School~~

CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope.

## ~~3 Atoms: The Building Blocks of Matter~~

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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope.

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Modern Chemistry 105 Chapter Test Name Class Date Chapter Test A, continued Use this figure to answer questions 7 and 8. \_\_\_\_\_ 7. A solution containing 35 g of  $Li_2SO_4$  dissolved in 100 g of water is heated from  $10^\circ C$  to  $90^\circ C$ . According to information in the figure, this temperature change would result in a. an additional 5 g of  $Li_2SO_4$  in ...

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